

## New phosphorus ylides in reactions of tertiary phosphines with phosphorylated quinone methide

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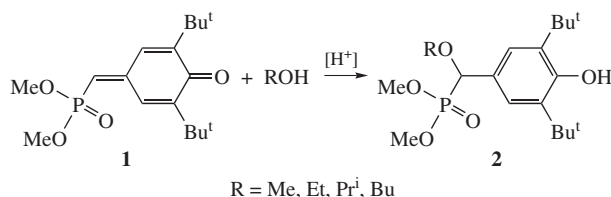
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DOI: 10.1016/j.mencom.2009.01.015

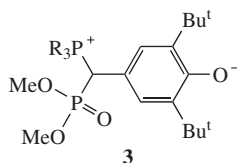
A series of new phosphonium ylides have been synthesized by the reaction of dimethyl-(3,5-di-*tert*-butyl-4-oxo-2,5-cyclohexadienylidene)methylphosphonate with tertiary phosphines, the structure of dimethoxyphosphoryl-3,5-di-*tert*-butyl-4-hydroxyphenylmethylidene triphenylphosphorane has been confirmed by X-ray single crystal diffraction.

The reactions of phosphorylated quinone methides with nucleophilic reagents result in formation of more thermodynamically stable aromatic phenol systems. Such reactions proceed, as a rule, by the 1,6-addition mechanism.

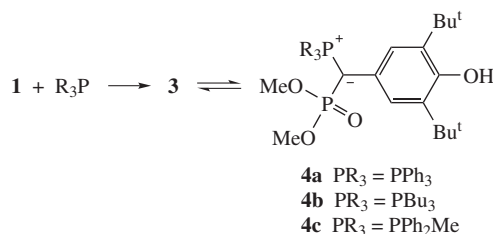
For example, dimethyl-(3,5-di-*tert*-butyl-4-oxo-2,5-cyclohexadienylidene)methylphosphonate **1** reacts with alcohols in the presence of catalytic amounts of mineral acids to give corresponding phenols **2**.<sup>1</sup>



Phosphorus-containing nucleophiles were not previously examined in such reactions. Here, we report the reactions of phosphorylated quinone methide **1** with tertiary phosphines, in particular, triphenyl-, tributyl- and methyldiphenyl phosphines, carried out in order to obtain phenolate phosphobetaines **3**.



Phosphorylated quinone methide **1** with tertiary phosphines gives stable colourless crystalline products with high melting points. The structure of all products was proved by spectral methods and elemental analysis.<sup>†</sup> The absorption band of the quinone carbonyl group (1632 cm<sup>-1</sup>) disappears in the IR spectra while the absorption band of the hydroxyl groups at 3150–3250 cm<sup>-1</sup> of the crystalline products appears, the frequency of the phosphoryl group absorption band being shifted from 1248 to 1145–1150 cm<sup>-1</sup>. The <sup>31</sup>P NMR spectra have two doublets of phosphonate and phosphonium phosphorus atoms due to their spin–spin coupling, observed at 16–22 and 28–34 ppm, respectively. Thus, spectral data give evidence for the formation of phosphonium ylides **4** rather than isomeric phenolate betaines **3**. The following scheme may be proposed with the C–H hydrogen atom of the quinone methide migrating in the course of the reaction towards oxygen atom of the phenolate anion.



The IR spectra of the products dissolved in acetonitrile reveal the sharp decrease of the intensity of the hydroxyl absorption band, thus giving evidence for the predominance of tautomeric form **3** in solution. As it was shown by X-ray single crystal diffraction (Figure 1),<sup>‡</sup> ylide tautomeric form is stabilized by strong intermolecular [O–H...O=P (x, y, z + 1)] hydrogen bonds, which lead to the formation of infinite chains along the *c*-axis: O–H 0.83(4), H...O=P 1.87(4), O...O=P 2.678(5) Å, ∠O–H...O 166.5°.

The carbon atom bearing a negative charge has a nearly planar configuration with the sum of valence angles of 359.92°. Typical of *sp*<sup>2</sup>-hybridized carbon atom the P(3)–C(1) bond is eclipsed

<sup>†</sup> Dimethyl-(3,5-di-*tert*-butyl-4-oxo-2,5-cyclohexadienylidene)methyl-phosphonate **1**: yield 76%, mp 105–107 °C. IR (ν/cm<sup>-1</sup>): 1632 (C=O), 1584 (C=CH), 1248 (P=O), 1056, 1036 (POC). <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ: 18.26. <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 1.32 (s, 9H, CMe<sub>3</sub>), 1.34 (s, 9H, CMe<sub>3</sub>), 3.82 (s, 3H, MeO), 3.85 (s, 3H, MeO), 6.04 (m, 1H, PCH), 6.82 (s, 1H, CH=C–Bu<sup>t</sup>), 7.99 (s, 1H, CH=C–Bu<sup>t</sup>). Found (%): C, 62.01; H, 8.19; P, 9.72. Calc. for C<sub>17</sub>H<sub>27</sub>O<sub>4</sub>P (%): C, 62.58; H, 8.28; P, 9.51.

Dimethoxyphosphoryl-3,5-di-*tert*-butyl-4-hydroxyphenylmethylidene triphenylphosphorane **4a**: yield 67%, mp 128.7 °C. IR (ν/cm<sup>-1</sup>): 1145 (P=O), 3200 (OH). <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ: 18.04 (d, <sup>2</sup>J<sub>P-P</sub> 76.29 Hz), 27.91 (d, <sup>2</sup>J<sub>P-P</sub> 76.29 Hz). Found (%): C, 71.43; H, 7.13. Calc. for C<sub>35</sub>H<sub>42</sub>O<sub>4</sub>P<sub>2</sub> (%): C, 71.43; H, 7.14.

Dimethoxyphosphoryl-3,5-di-*tert*-butyl-4-hydroxyphenylmethylidene tributylphosphorane **4b**: yield 85%, mp 164 °C. IR (ν/cm<sup>-1</sup>): 1148 (P=O), 3150 (OH). <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ: 22.65 (d, <sup>2</sup>J<sub>P-P</sub> 85.45 Hz), 34.11 (d, <sup>2</sup>J<sub>P-P</sub> 85.45 Hz). Found (%): C, 65.97; H, 10.23. Calc. for C<sub>29</sub>H<sub>54</sub>O<sub>4</sub>P<sub>2</sub> (%): C, 65.90; H, 10.34.

Dimethoxyphosphoryl-3,5-di-*tert*-butyl-4-hydroxyphenylmethylidene methyldiphenylphosphorane **4c**: yield 92%, mp 155.4 °C. IR (ν/cm<sup>-1</sup>): 1150 (P=O), 3250 (OH). <sup>31</sup>P NMR (CDCl<sub>3</sub>) δ: 16.98 (d, <sup>2</sup>J<sub>P-P</sub> 80.00 Hz), 34.09 (d, <sup>2</sup>J<sub>P-P</sub> 80.00 Hz). Found (%): C, 68.52; H, 7.73. Calc. for C<sub>30</sub>H<sub>38</sub>O<sub>4</sub>P<sub>2</sub> (%): C, 68.44; H, 7.60.